Creation Date : 2022/10/3
Last Revision Date : 2023/7/7

SAFETY DATA SHEET (SDS)

1. IDENTIFICATION

Product name ALA GARDEN Farm

Company Seiwa Fertilizer Ind. Co.,Ltd.

Address 3-4, 4 Chome, Bingo Machi, Chuo-ku, Osaka City

 Telephone
 +81-6-6231-3771

 FAX
 +81-6-6231-1988

 Emergency telephone number
 +81-6-6231-3771

 Office hour
 Mon-Fri 9:00-17:00

Recommended uses and restrictions on use Fertilizer

2. HAZARDS IDENTIFICATION

GHS classification of the substance or mixture

Physical hazardsNot classifiedHealth hazardsNot classifiedEnvironmental hazardsNot classified

Label elements

Pictograms or hazard symbolsNoneSignal wordNoneHazard statementsNonePrecautionary statements (Prevention)None

(Response) None(Storage) None

(**Disposal**) Observe all federal, state and local regulations when

disposing of the substance and container.

Others No information available

Other hazards Irritation to eye, respiratory system, and skin

3. COMPOSITION / INFORMATION ON INGREDIENTS

Substance/mixture Mixture

Chemical Name Compound fertilizer

Chemical Name	ENCS No.	CAS RN	Percent
Ammonium Sulfate	1-400	7783-20-2	48-51%
Single Superphosphate	not-disclosed	8011-76-5	10-30%
Triple Super Phosphate	not-disclosed	65996-95-4	15-25%
Potassium Chloride	1-228	7447-40-7	17-18%
Citric Acid, Anhydrous	2-1318	77-92-9	10%Max

Diatomaceous Earth (Uncalcioed)	1-548	69012-64-2	1%Max
Others			1%Max

Components	ENCS No.	CAS RN	Guaranteed component (%)
Total nitrogen	not-disclosed	not-disclosed	10.0%
Ammoniac nitrogen as N	not-disclosed	not-disclosed	10.0%
Soluble phosphorus as P ₂ O ₅	not-disclosed	not-disclosed	10.0%
Water solbule phosphorus as P ₂ O ₅	not-disclosed	not-disclosed	8.0%
Water soluble potassium as K ₂ O	not-disclosed	not-disclosed	10.0%

4. FIRST-AID MEASURES

Inhalation Remove victim to fresh air and keep at rest in a position

comfortable for breathing. Get medical advice if needed.

Skin contact Wash off immediately with soap and plenty of water. If

skin irritation or rash occurs, get medical

advice/attention.

Eye contact Rinse cautiously with water for more than 15 minutes.

> Remove contact lenses, if present and easy to do. Continue rinsing. If eye irritation persists, get medical

advice/ attention.

Ingestion Rinse mouth with clean water and get medical attention.

> Do not force to vomit. If vomiting occurs, the head should be lowered to prevent vomiting from entering the

lungs.

5. FIRE-FIGHTING MEASURES

Suitable extinguishing media Water spray, carbon dioxide gas, dry chemical, foam

Unsuitable extinguishing media Straight stream water

The product is hard to burn. Thermal decomposition can Specific hazards arising from the chemical

product lead to release of ammonia gas, sulfur oxide.

Special extinguishing method Fire-extinguishing work is done from the windward.

> Entry to non-involved personnel should be controlled around the area. Remove movable containers if safe to do so. Cool the container fully after extinguishing the fire.

Special protective actions for fire-fighters Fire-extinguishing work is done from the windward and

avoid inhaling harmful gases. Wear respiratory ptotection

according to the situation.

6. ACCIDENTAL RELEASE MEASURES

emergency procedures

Personal precautions, protective equipment and See Section 8. Wear appropriate personal protective equipment to avoid adhering it on skin or in eyes, or

inhaling dust.

Environmental precautions

Do not let product enter drains and water course. Do not discharge to the environment due to containing ingredients that cause eutrophication of the water system.

Methods and materials for contaminent and methods and materials for cleaning up

Sweep up or vacuum scattered particles in order to prevent the dispersion of dust, and collect them in an empty container. Keep it away from drains.

Secondary disaster prevention measures

No information available

7. HANDLING AND STORAGE

Handling Technical measures See Section 8. Wear appropriate personal protective

equipment. Use in a well-ventilated area. Use a ventilation, local exhaust according to the situation.

Safety handling precautions Keep out of contact with alkaline substances, oxidizing

agents, and high temperature substances. Wash hands and face after handling. Prevent dispersion of dust. Be

careful not to inhale or ingest dust. Take off

contaminated clothes. Wash them before reusing. No smoking, or eating and drinking when handling. Use effective amount of the product as fertilizer, which is undiluted or diluted with an appropriate amount of water,

according to the crop.

Storage Storage conditions Store away from sunlight, in a cool, dark, and dry place.

Use a sealable container without damage and leakage. Keep container tightly closed. Leaving it open may cause consolidation due to moisture absorption. Moisture absorption may accelerate the decomposition and invalidation of components. Lighting and ventilation equipment necessary for storing or handling dangerous and harmful substances will be installed in the storage

location.

Incompatible substances Oxidizing agents, reducing agents, alkaline substances,

high temperature substances

Safe packaging material Sealable container without damage and leakage.

Polyethylene, polypropylene.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Control parameters

Not set up

Exposure limits Japan Society for

Occupational Health (JSOH) (2022)

Dust Type 3 Other inorganic and organic acid Inhalable dust: 2mg/m³, Total dust: 8mg/m³

Engineering controls

Install a closed system or local exhaust in the workplace where dust is generated. Implement regulatory management and equipment management to reduce exposure.

Personal protective equipment

Respiratory protection Dust respirator

Hand protection Rubber or PVC protective gloves

Eye protection Protective glasses - with side plates, or goggles **Skin and body protection** Protective long sleeve clothing, protective apron

General hygiene Wash hands and face after handling.

considerations No smoking, or eating and drinking when handling.

Take off contaminated clothes. Wash them before

reusing.

9. PHYSICAL AND CHEMICAL PROPERTIES

Appearance Physical state Solid

Form Granular
Color Grayish white

Odour Slightly sour odor

pH 3-3.5 (1% water solution)

Ignition point None

Flammability Nonflammable

Flammability or explosive limits Min: None Max: None

Bulk specific gravity 0.95 - 1.10

Solubilities Water No data available

Other solvents No data available

Other physicochemical properties No data available

10. STABILITY AND REACTIVITY

Reactivity, chemical stabilityStable under normal temperatures and pressures.

Possibility of hazardous reactionsNot reacted under normal handling conditions.

Conditions to avoid Avoid contact with, mixing with, or storage close to

water and incompatible materials.

Avoid direct sunlight, and store in cool, dark, and dry

place.

Incompatible materials Oxidizing agents, reducing agents, alkaline substances,

high temperature substances.

Hazardous decomposition productsThermal decomposition can lead to release of ammonia

gas, sulfur oxide.

11. TOXICOLOGICAL INFORMATION

Product hazard information No data available

Components hazard information

Ammonium Sulfate
Acute Toxicity

Oral Human TDLo = 1,500 mg/kg

Rat LD50 = 2,000 - 4,250 mg/kgMouse LD50 = 640 mg/kg

Livestock LDLo = 3,500 mg/kg

Intraperitoneal Mouse LD50 = 610 mg/kg

Skin Mouse / Rat LD50 > 2,000 mg/kg

Inhalation Rat LD50 (8h) $> 1,000 \text{ mg/m}^3$

Chronic toxicity Rat NOAEL = 886 mg/kg/day (mixed feeding)

Skin corrosion / irritationRabbit No irritationSerious eye damage/irritaionRabbit No irritationSkin sensitizationNo data availableRespiratory sensitizationNo data availableGerm cell mutagenicityAmes test; Negative

Chromosome aberration test; Negative

Reproductive toxicity CarcinogenicityNo data available
No data available

STOT-single exposure Oral exposure caused staggering, blunt hemp, and effort

breathing in rats, and mydriasis, irregular breathing, and convulsions that spread from local areas (face and extremities) to the whole body in rabbits, died of cardiac arrest as a result. However, no histopathological changes were observed in each organ of rats and rabbits. On the other hand, EEG examination revealed rabbits to be a classic case of hyperammonemia. Based on these results, the post-exposure effects were speculated to be ammonia neurotoxicity. Since it exceeds the upper limit of the guidance value range (2,000 mg/kg) in rats and is found at 1,500 mg/kg in rabbits, it may cause damage to the

nervous system.

In humans, inhalation exposure to 0.1-0.5~mg ammonium sulfate/m³ aerosol for two to four hours produced no pulmonary effects. At 1 mg ammonium sulfate/m³ very slight pulmonary effects in the form of a decrease in expiratory flow, in pulmonary flow resistance and dynamic lung compliance were found in healthy

volunteers after acute exposure.

STOT-repeated exposure Rat inhalation toxicity test (0.3 mg/L, 8h/day, 14days);

No effect

Rat 13-week repeated oral dose study (1,792 mg/kg/day)

; No effect

Single Superphosphate

Acute Toxicity

Oral No reliable information

SkinNo data availableInhalationNo data availableChronic toxicityNo data available

Skin corrosion / irritation Although there is no data, there is a description that it is

considered to cause mild skin irritation.

Serious eye damage / irritation There is a statement that it is considered to be irritating if

it gets in the eyes.

Skin sensitization No data available

Respiratory sensitizationNo data availableGerm cell mutagenicityNo data availableReproductive toxicityNo data availableCarcinogenicityNo data availableSTOT-single exposureNo data availableSTOT-repeated exposureNo data available

Other relevant information Excess inhalation of dust may cause irritation of the

nose, throat and respiratory tract. Prolonged and

repeated contact may cause mild irritation to the skin. In contact with the eyes, dust may cause irritation, redness and abrasion. Ingestion of large amounts of dust may cause gastrointestinal upset and abdominal pain. (OECD

SIDS)

Triple Super Phosphate

Acute Toxicity

Oral No data available
Skin No data available
Inhalation No data available

Repeated dose toxicity Rat, Gavage, Doses; 0, 250, 750, and 1,500 mg/kg/day

The stomach submucosal effect seen at all doses may have been a result of irritation due to the low pH (2-3) of

the test solution that was gavaged.

General toxicity: LOAEL = 250 mg/kg/day

Reproduction/developmental toxicity : LOAEL = 1,500

mg/kg/day

Skin corrosion / irritationMild irritationSerious eye damage / irritaionMild irritationSkin sensitizationNo data availableRespiratory sensitizationNo data available

Mammalian chromosome aberration

test

Chinese hamster ovary cells, Negative

Germ cell mutagenicity No data available

Reproductive toxicity Rat, Gavage, 250-1,500 mg/kg/day, Exposure period up

to 28 days for males and up to 53 days for females, No obvious signs of clinical toxicity, Mating performance

and fertility were not affected

CarcinogenicityNo data availableSTOT-single exposureNo data availableSTOT-repeated exposureNo data available

Potassium Chloride

Acute Toxicity

Oral Rat : LD50 = 3,020 mg/kg

Rat : LD50 = 2,600 mg/kgRat : LDL0 = 2,430 mg/kg

Guinea pig : LDL0 = 2,500 mg/kg

Intraperitoneal administration Rat : LD50 = 660 mg/kg

Rat : LDL0 = 825 mg/kg

Guinea pig: LDL0 = 900 mg/kg

Intravenous injection Rat : LD50 = 142 mg/kg

Rat: LDL0 = 117 mg/kg

Guinea pig : LDL0 = 77 mg/kg Guinea pig : LDL0 = 130 mg/kg

Intra-arterial injection Guinea pig: LDL0 = 130 mg/kg **Intracardiac administration** Guinea pig: LDL0 = 40 mg/kg

Inhalation No data available

Repeated dose toxicity

Oral Human: NOAEL > 80 mmol (approx. 85mg)

KCl/kg/day

Rat: NOAEL > 1,820 mg/kg/day

Inhalation No data available

Hazardous to human

Inhalation Cough, Sore throat (ICSC)

Skin No description of symptoms (ICSC)

Eyes Redness, Pain (ICSC)

Oral intake Abdominal pain, Diarrhea, Rausea, Vomiting, Weakness,

cramps (ICSC)

Skin irritation A threshold concentration for skin irritancy of 60 % was

seen when potassium chloride in aqueous solution was in

contact with skin of human. The

threshold concentration when applied to broken skin was

5 %.

Mutagenicity No gene mutations were reported in bacterial tests, with

and without metabolic activation. However, high

concentrations of KCl showed positive results in a range of genotoxic screening assays using mammalian cells in culture. The action of KCl in culture seems to be an indirect effect associated with an increased osmotic pressure and cocentration. Therefore KCl, do not have

any direct revance in the intact body were such

cocentrations can not occur. Further studies using in vivo systems are not considered necessary under SIDS.

Germ cell mutagenicity A developmental study revealed no foetotoxic or

tertogenic effect of KCl in doses up to 235 mg/kg/day

(mice) and 310 mg/kg/day (rats).

Reproductive toxicityNo data available

Carcinogenicity No evidence of treatment-related carcinogenicity was

observed in rats administered up to 1820 mg KCl/kg body weight/day through the food in a 2 year study.

STOT-single exposure No data available STOT-repeated exposure No data available

Citric Acid, Anhydrous

Acute Toxicity

Oral Mouse: LD50 = 5,040 mg/kg

Rat : LD50 = 3,000-12,000 mg/kg

Rabbit : Lethal dose = 7,000 mg/kg (probably lowest

lethal dose)

Intraperitoneal Mouse: LD50 = 903 mg/kg

Rat: LD50 = 290 mg/kg

Intravenous Mouse: LD50 = 42 mg/kg

Rabbit: LD50 = 330 mg/kg

Subcutaneous Mouse: LD50 = 2,700 mg/kg

Rat : LD50 = 5,500 mg/kg

Inhalation Human: Cough. Shorthness of breath. Sore throat.

Chronic toxicity No data available

Skin corrosion / irritation Rabbit : 500 mg/24h (mild)

Human: Redness.

Serious eye damage / irritaion Rabbit : 0.75 mg/24h (severe)

Human: Redness. Pain.

Respiratory sensitizationNo data availableSkin sensitizationNo data availableMutagenicity / GenotoxicityNo data availableCarcinogenicityNo data availableReproductive and DevelopmentalNo data available

toxicity

STOT-single exposure The substance is irritating to the eyes, skin, and

respiratory tract.

STOT-repeated exposure May cause tooth acid erosion.

Amorphous Silica (Diatomaceous Earth, Uncalcioed)

Acute Toxicity

OralNo data availableInhalationNo data availableSkinNo data available

Skin corrosion / irritation Prolonged contact may cause allergic dermatitis.

Serious eye damage / irritaion Cause pain

Respiratory sensitizationNo data availableSkin sensitizationNo data availableGerm cell mutagenicityNo data availableReproductive toxicityNo data available

STOT-single exposure May cause respiratory irritation. (respiratory tract

irritation)

STOT-repeated exposure Respiratory damage from prolonged or repeated

exposure.

In humans, chronic or repeated inhalation exposures have been reported to cause relapsing fevers like metal fume fever, but spontaneous reversal of pulmonary changes

have been reported. (ACGIH, 7th, 2001)

In a 12- to 18-month inhalation exposure study at 15 mg/m3 in rats, guinea pigs, and monkeys, increased aggregates of mononuclear cells and reticular fibrosis in lungs were observed in all animal species. (DFGOT vol.2

(1991))

12. ECOLOGICAL INFORMATION

Product hazardous information

No data available

Components hazardous information

Ammonium Sulfate

Hazardous to the aquatic environment (Acute)

Fish (Oncorhynchus mykiss, juvenile)

LC50 (96 h) = 173 mg/L

Fish (*Danio rerio*) LC50 (96 h) = 420mg/L Fish (Guppy) LC50 (96 h) = 126 mg/L

Invertebrate (Helisoma trivolyis)

LC50 (24 h) = 393 mg/L Crustacea (*Daphnia magna*) EC50 (96 h) > 100 mg/L Algae (Chlorella vulgaris) EC50 (18 days) = 2,700 mg/L

Perna viridis EC50 (96 h) = 47.7 mg/L Fish (Oncorhynchus gorbuscha, juvenile)

Hazardous to the aquatic environment (Long-term)

NOEC = 11 mg/L (61 days)

Readily degradable

Persistence and degradability

Hazardous to the ozone layer Not listed in the an

Not listed in the annex to the Montreal Protocol

Single Superphosphate

Hazardous to the aquatic

environment

Fish

Labeo rohita: LC50 (96h) = 3,460 mg/L Catla catla): LC50 (96h) = 2,620 mg/L Cirrhinus mrigala: LC50 (96h) = 1,560 mg/L Cyprinus carpio: LC50 (96h) = 3,900 mg/L Tilapia mossambica: LC50 (96h) = 5,900 mg/L

Daphnia (Daphnia carinata) EC0 (22.7C, 72h) = 68 mg/L EC50 (22.7C, 72h) = 80 mg/L EC95 (22.7C, 72h) = 5,115 mg/L EC100 (22.7C, 72h) = 5,800 mg/L EC0 (30.8C, 72h) = 75 mg/L EC5 (30.8C, 72h) = 82 mg/L EC50 (30.8C, 72h) = 1,825 mg/L

EC95 (30.8C, 72h) = 5,325 mg/L EC100 (30.8C, 72h) = 5.780 mg/L

Zooplankton

Moina micrura : Avg. LC50 (72h) = 1,625 mg/L Cyclops viridis : Avg. LC50 (72h) = 2,305 mg/L

Aquatic insect

Branchiura sowerbyi (worm) : Avg. LC50 (96h) = 3,320 mg/L

Chironomus (larvae):

Avg. LC50 (96h) = 1,510 mg/L

Dragonfly (larvae): Avg. LC50 (96h) = 1,133 mg/L

Mollusc

Planobis exustus : Avg. LC50 (96h) = 5,005 mg/LLymnaea leuteola : Avg. LC50 (96h) = 2,950 mg/L

Viviparus bengalensis:

Persistence and degradability No data available

BioaccumulationNo data availableMobility in soilNo data available

Triple Super Phosphate

Hazardous to the aquatic

environment

No data available

Persistence and degradabilityNo data availableBioaccumulationNo data availableMobility in soilNo data available

Potassium Chloride

All the studies compiled on the acute and chronic aquatic toxicity were > 100 mg/L. Thus it is concluded that KCl is not hazardous to freshwater organisms. Taking into considerations the background concentrations of KCl in seawater (K⁺ 380 mg/l and Cl⁻ 19,000 mg/l), it is concluded that there is no reason for further investigations of KCl on marine species. The low concern for the environment is supported by the absence of a bioaccumulation potential for the substance.

(OECD)

Fish

Hazardous to the aquatic environment (Acute)

Pimephales promelas : LC50 (24h) = 950 mg/L LC50 (48h) = 910 mg/L, LC50 (96h) = 880 mg/L Ictahurus punctatus : LC50 (48h) = 720 mg/L

Crustacea

Daphnia magna:

EC50 (24h) = 740 mg/L, EC50 (48h) = 660 mg/L, EC50 (48h) = 177 mg/L, EC50 (48h) = 141 mg/L *Ceriodaphnia dubia* : EC50 (48h) = 630 mg/L

Diaton

Crustacea

Nitzschia linearis: EC50 (120h) = 1,337mg/L

Hazardous to the aquatic environment (Long-term)

Daphnia magna: EC50 (21 d) = 130 mg/L,

LOEC (21d) = 101 mg/L (16% reproduction impairment)

Citric Acid, Anhydrous

Hazardous to the aquatic environment, Short term

Fish

Lepomis macrochirus (Bluegill):

LC50 (96h) = 1,516 mg/L Leuciscus idus (Cyprinidae) :

LC50 (96h) = 440-760 mg/L (not neutralised)

Crustacea

Daphnia magna: EC0 = 1,206 mg/L,

EC50 = 1,535 mg/L,

EC100 = 2,083 m/L (neutralised)

Daphnia magna:

EC0 = 73 mg/L, EC50 = 85 mg/L, EC100 = 98 mg/L (not neutralised) Carcinus maenas (European green crab):

LC50 (48h) = 160 mg/L

Algae

Scenedesmus quadricauda : ECO (7d) = 640 mg/L Pavlova lutheri :

TLC (7d) = 1-300 mg/L "saltwater"

Bacteria

Microcystis aeruginosa: EC0 (8d) = 80 mg/L

Nitrosomonas sp.:

No inhibition on NH3 oxidation at 100 mg/L Pseudomonas putida : EC0 (16h) > 10,000 mg/L

Uronema parduzci : TLC = 622 mg/L

Hazardous to the aquatic Fish

environment, Long term *Carassius auratus* :

LC0 = 625 mg/L, LC100 = 894 mg/L "long-time exposure in soft water"

Crustacea

Daphnia magna:

EC0 = 80 mg/L, EC100 = 120 mg/L "long-time exposure in soft water"

Persistence and degradability Readily degradable

Hazardous to the ozone layerNot listed in the annex to the Montreal Protocol.

Amorphous Silica (Diatomaceous Earth, Uncalcioed)

Ecotoxicity No ecotoxicity under 67/548/EEC (DSD) and regulation

(EC) No. 1272/2008 (CLP).

Mobility No mobility under normal circumstances.

Degradability It does not decompose inorganic matter.

Hazardous to the ozone layerNot listed in the annex to the Montreal Protocol.

13. DISPOSAL CONSIDERATIONS

Waste from residues Follow the relevant laws and local disposal regulations.

Entrust disposal to and industrial waste contractor or local public body that is authorized by the prefectual

governonr where available.

Store waste in appropreate condition and do not drain

into watercourse.

Contaminated container and contaminated

packaging

Either clean and recycle the containers, or dispose of them suitably according to the relevant laws and regulations, and local disposal regulations. When

disposing of empty containers, make sure to discard the

contents completely.

14. TRANSPORT INFORMATION

International regulations

ADR/RID

UN number Not regulated Proper shipping name Not regulated

UN classification Not regulated Packing group Not regulated

IMDG

UN number Not regulated
Proper shipping name Not regulated
UN classification Not regulated
Packing group Not regulated
Marine pollutant (Sea) Not applicable
the IBC code Not regulated

ICAO/IATA

UN number Not regulated
Proper shipping name Not regulated
UN classfication Not regulated
Packing group Not regulated

Japanese regulations

Information on road transport Not regulated

regulation

Information on marine transportNot regulated

regulation

Marine pollutant substanceNot regulatedInformation on air transport regulationNot regulated

Emergency Response Guidebook Not regulated

(Yellow-card)

Special precautions Check the container is not damaged, corroded, or leaked

before transported. Avoid direct sunlight. Be careful not to fall, drop, or damage when loading, and ensure not to collapse. Equip the truck and ship with protective equipments (gloves, glasses, masks, etc.), and fire extinguishers, tools necessary for emergency.

15. JAPANESE REGULATORY INFORMATION

Fertilizer Regulation Act Fertilizer - Compound fertilizer

Industrial Safety and Health Act

Dangerous and hazardous substances subject to notice

and package the name: Not listed

Water Pollution Prevention Act Harmful substances (Cabinet Order Article 2, 26):

Ammonia, Ammonium compounds, Nitrite compounds

and Nitric acid compounds

Emission standard: 100mg/L(total of ammonium nitrogen

x 0.4, nitrite nitrogen, and nitrate nitrogen)

Act on Prevention of Marine Pollution

and Maritime Disaster

Hazardous liquid substance (Z class substance):

Ammonium Sulfate solution, Potassium chloride solution

Air Pollution Control Act

Chemical Substance Emission Control

Promotion Act (PRTR)

Not applicable Not applicable

Foreign Exchange and Foreign Trade Act

Export Trade Control Order row 16 of appended table 1 [HS: 3102 Mineral or chemical fertilisers, nitrogenous]

16. OTHER INFORMATION

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- 4) Ministry of Health, Labour and Welfare, Shokuba no Anzen Site, JAPAN: Substances subject to mandatory labeling and SDS delivery based on the Industrial Safety and Health Act -Japanese, https://anzeninfo.mhlw.go.jp/anzen/gmsds/gmsds640.html
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- 8) un-no-un-number.com : UN Number -Japanese, https://www.un-no-un-number.com/index.html
- 9) Institute of Chemicals and Regulation : Act on Prevention of Marine Pollution and Maritime Disaster, Hazardous Liquid Substance List Japanese, https://www.chemicalsubstance.com/kaiyoosenboushiho/zrui.html

Ammmonium sulfate, references

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- 11) SIDS INITIAL ASSESSMENT PROFILE, Ammonium sulfate, https://hpvchemicals.oecd.org/UI/handler.axd?id=4e17215a-1849-4517-91a4-a31ba0c28c46
- 12) NIPPON STEEL Chemical & Material Co., Ltd. (2012) : Ammonium Salfate, SDS -Japanese version

Single Superphosphate, references

- 13) Incorporated Administrative Agency National Institute of Technology and Evaluation (NITE), JAPAN: NITE Chemical Risk Infomation Plattform (NITE-CHRIP) -English version, Single Superphosphate (SSP), https://www.nite.go.jp/en/chem/chrip/chrip_search/cmpInfDsp?cid=C005-861-
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